

! (H P0↔)	" 85.0%	" 85.0%	86.0%
# \$%&' ()	* 25	* 25	+ 25
, - . (/ H+O)%mmol /g	* 0.0002	* 0.0002	+ 0.0002
1 2 3 (Cl)	* 0.0003%	* 0.0005%	+ 0.0003%
4. 5 (S0↔)			